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A HIGHLY DIASTEREOSELECTIVE OXIDATION OF A CROWDED GEM-DISULPHIDE

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A HIGHLY DIASTEREOSELECTIVE OXIDATION OF A CROWDED *GEM*-DISULPHIDE

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In all of the kinetically controlled oxidations to *gem*-disulphoxides, the *meso*-epimer predominated. Spontaneous interconversion of epimeric $(\text{PhCH}_2\text{SO})_2\text{CPh}_2$ is made possible by the lower C-S bond energies in disulphoxide, permitting a room temperature homolytic dissociation-recombination mechanism to operate, being an example of kinetic and equilibrium control (C. Y. Meyers, L. L. Ho, A. Ohno and M. Kagami, *Tetrahedron Letters* 1974, 729).

We have now found an unexpected diastereoselective oxidation of another crowded *gem*-disulphide $(\text{PhCH}_2)_2\text{C}(\text{SCH}_3)_2$ (1) which afforded the *gem*-disulphoxide 2 as sole product. The NMR studies of 2 and its constitutional isomer, the monosulphone 3, showed amazing shielding features which suggested an intramolecular interaction. *Syn*-elimination reactions gave *Z*-olefins as sole products. When 2 is treated in boiling chloroform, fragmentation takes place and the corresponding *Z*-methylsulphinyl-alkene 4 is formed. Thermolysis of 2 in presence of BF_3 proceeded partly through bimolecular disproportionation (giving the corresponding monosulphoxide and sulphonyl-sulphoxide), followed by elimination of sulphenic acid and affording *Z*-methylsulphinyl-alkene 4, *Z*-methylsulphonyl-alkene 5, and a mixture of *E*- and *Z*-methylthio-alkenes 6a,b. All these reactions were highly stereospecific and gave 80% and higher yields of pure diastereoisomers. A detailed investigation of these unexpected selectivities is in progress.