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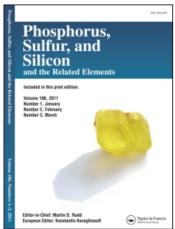
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A HIGHLY DIASTEREOSELECTIVE OXIDATION OF A CROWDED GEM-DISULPHIDE

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A HIGHLY DIASTEREOSELECTIVE OXIDATION OF A CROWDED GEM-DISULPHIDE

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In all of the kinetically controlled oxidations to gem-disulphoxides, the meso-epimer predominated. Spontaneous interconversion of epimeric (PhCH₂SO)₂CPh₂ is made possible by the lower C-S bond energies in disulphoxide, permitting a room temperature homolytic dissociation-recombination mechanism to operate, being an example of kinetic and equilibrium control (C. Y. Meyers, L. L. Ho, A. Ohno and M. Kagami, Tetrahedron Letters 1974,729).

We have now found an unexpected diastereoselective oxidation of another crowded gem-disulphide (PhCH₂)₂ C(SCH₃)₂ ($\underline{1}$) which afforded the gem-disulphoxide 2 as sole product. The NMR studies of 2 and its constitutional isomer, the monosulphone 3, showed amazing shielding features which suggested an intramolecular interaction. Syn-elimination reactions gave Z-olefins as sole products. When 2 is treated in boiling chloroform, fragmentation takes place and the corresponding Z-methylsulphinyl-alkene 4 is formed. Thermolysis of 2 in presence of BF, proceeded partly through bimolecular disproportionation (giving the corresponding monosulphoxide and sulphonyl-sulphoxide), followed by elimination of sulphenic acid and affording Z-methylsulphinyl-alkene 4, Z-methylsulphonyl-alkene 5, and a mixture of E- and Z-methylthio-alkenes 6a,b. All these reactions were highly stereospecific and gave 80% and higher yields of pure diastereoisomers. A detailed investigation of these unexpected selectivities is in progress.